

An N-unprotected Fischer pyrrolidinylidene complex

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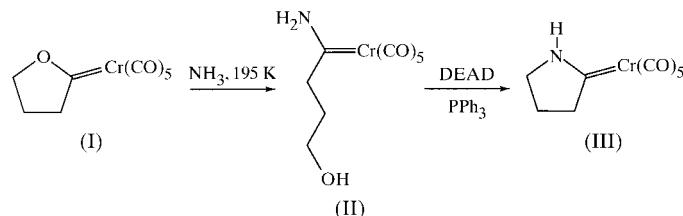
The synthesis and crystal structure of pentacarbonyl(pyrrolidin-2-ylidene)chromium(0), $[\text{Cr}(\text{C}_4\text{H}_7\text{N})(\text{CO})_5]$, is reported. The compound shows strong interaction between the lone pair at nitrogen and the carbene C atom, and weak intermolecular hydrogen bonds.

Comment

Fischer aminocarbene complexes are versatile reagents for stereoselective organic synthesis (Dötz, 1984; Hegedus, 1995; Wulff, 1995; Barluenga, 1996). Cycloalkylidene complexes are potential precursors for spiro-cycloaddition and annelation reactions. Oxacyclopentylidene complexes are readily accessible by cyclization of alkynols at the metal template (Dötz *et al.*, 1987) and can be transformed into azacyclopentylidene (pyrrolidinylidene) complexes by an efficient one-pot procedure recently described by us (Haase & Dötz, 1999). The protocol is characterized by smooth conditions and provides the first generally applicable (Haase *et al.*, 1999; Dötz *et al.*, 1997) access to N-unprotected azacyclopentylidene complexes.

The structure of (III) was determined by X-ray crystallography to study the ring conformation and bond lengths in the absence of steric hindrance induced by substituents at the N atom. Complex (III) shows a ³E conformation, with an angle of 11.3 (3)° between the C4/N1/C1/C2 and C4/C3/C2 planes, C3 lying 0.181 (4) Å above the C4/N1/C1/C2 plane. Characteristic details are the planarity at the N and the carbene C atom which indicates the strong π -interaction between these atoms, supported by the very short nitrogen–carbene bond of 1.296 (2) Å and the minor distortions from an exact trigonal environment of the Nsp^2 atom [C1–N1–H1 121.7 (14)°]. In contrast, known N-methyl-substituted azacyclopentylidene complexes show major steric interactions between the pentacarbonylchromium fragment and the N-methyl group resulting in C1–N1–NCH₃ angles of 125.8 (Dötz *et al.*, 1997)

or 127.7° (Dötz *et al.*, 1999) and longer nitrogen–carbene bonds (1.313 and 1.3059 Å, respectively). Complex (III) shows almost ideal octahedral coordination around the Cr atom. The Cr–C distance of the *trans*-carbonyl group is between 0.014 and 0.035 Å shorter than to the *cis*-carbonyl ligands, as expected for Fischer carbene complexes (Fischer, 1974). The crystal structure is stabilized by weak intermolecular hydrogen bonds between the N1 H atom and the O1Eⁱ atom of a carbonyl ligand [N1–H1 0.86, H1···O1Eⁱ 2.26 (2), N1···O1Eⁱ 3.069 (2) Å and N1–H1···O1Eⁱ 157°; symmetry code: (i) $x + 1, y, z$].



Experimental

Pentacarbonyl(oxacyclopentylidene)chromium(0), (I), was subjected to ammonolysis and the intermediate formed, *i.e.* acyclic amino-carbene complex (II), was recycled under Mitsunobu conditions overnight. Recrystallization of the pure product from *n*-pentane/dichloromethane yielded complex (III) as yellow crystals in 65% overall yield.

Crystal data

$[\text{Cr}(\text{C}_4\text{H}_7\text{N})(\text{CO})_5]$
 $M_r = 261.16$
Orthorhombic, $P2_12_12_1$
 $a = 6.7152$ (3) Å
 $b = 10.1507$ (5) Å
 $c = 15.5447$ (9) Å
 $V = 1059.59$ (9) Å³
 $Z = 4$
 $D_x = 1.637$ Mg m⁻³

Mo $K\alpha$ radiation
Cell parameters from 14785 reflections
 $\theta = 3.3\text{--}27.3^\circ$
 $\mu = 1.083$ mm⁻¹
 $T = 123$ (2) K
Plate, yellow
0.40 × 0.10 × 0.05 mm

Data collection

Nonius KappaCCD diffractometer
Rotation in φ and ω , 2° scans
14 711 measured reflections
2630 independent reflections
2402 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$
 $\theta_{\text{max}} = 28.28^\circ$
 $h = -8 \rightarrow 8$
 $k = -13 \rightarrow 13$
 $l = -20 \rightarrow 20$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.024$
 $wR(F^2) = 0.057$
 $S = 1.017$
2630 reflections
148 parameters
H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0307P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.25$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.22$ e Å⁻³
Absolute structure: Flack (1983);
1097 Friedel pairs
Flack parameter = -0.015 (17)

All H atoms were located by a difference Fourier synthesis and refined with fixed individual displacement parameters [$U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C or N})$], using a riding model with C–H = 0.99 Å, while the coordinates of the H(N) atom were refined.

Data collection: COLLECT (Nonius, 1998); cell refinement: DENZO-SMN (Otwinowski & Minor, 1997); data reduction:

DENZO-SMN (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97*.

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References

- Barluenga, J. (1996). *Pure Appl. Chem.* **68**, 543–552.
- Dötz, K. H. (1984). *Angew. Chem.* **96**, 573–594; *Angew. Chem. Int. Ed. Engl.* **23**, 587–608.
- Dötz, K. H., Haase, W.-C., Klumpe, M. & Nieger, M. (1997). *Chem. Commun.* pp. 1217–1218.
- Dötz, K. H., Klumpe, M. & Nieger, M. (1999). *Chem. Eur. J.* **5**, 691–699.
- Dötz, K. H., Sturm, W. & Alt, H. G. (1987). *Organometallics*, **6**, 1424–1427.
- Fischer, E. O. (1974). *Angew. Chem.* **86**, 651–682.
- Flack, H. D. (1983). *Acta Cryst. A* **39**, 876–881.
- Haase, W.-C. & Dötz, K. H. (1999). *Tetrahedron Lett.* **40**, 2919–2920.
- Haase, W.-C., Nieger, M. & Dötz, K. H. (1999). *Chem. Eur. J.* **5**, 2014–2024.
- Hegedus, L. S. (1995). *Comprehensive Organometallic Chemistry II*, Vol. 12, edited by E. W. Abel, F. G. Stone & G. Wilkinson, pp. 549–576. Oxford: Pergamon Press.
- Nonius (1998). *COLLECT*. Nonius BV, Delft. The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods Enzymol.* **276**, 307–326.
- Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1990). *Acta Cryst. A* **46**, 467–473.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Wulff, W. D. (1995). *Comprehensive Organometallic Chemistry II*, Vol. 12, edited by E. W. Abel, F. G. Stone & G. Wilkinson, pp. 469–547. Oxford: Pergamon Press.